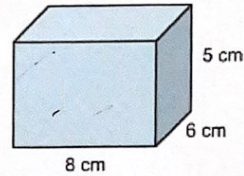




Quick Test Density

1. The diagram shows a solid cuboid made of wood.
The wood has a density of 0.56 grams per cm cubed.
Calculate the mass of the cuboid.



$$\text{Density} = \frac{\text{Mass}}{\text{Volume}}$$

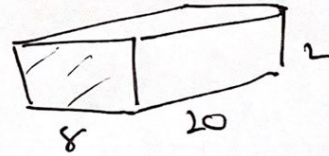
$$0.56 = \frac{\text{Mass}}{240}$$

$$0.56 \times 240 = \text{Mass}$$

$$\text{Mass} = \underline{\underline{134.4\text{g}}}$$

$$\begin{aligned}\text{Volume} &= \text{Area} \times \text{Depth} \\ &= (8 \times 6) \times 5 \\ &= 240\text{cm}^3\end{aligned}$$

2. The density of gold is 19.4 grams per cm cubed.
A gold ingot is produced in the shape of a cuboid.
The cuboid has length 20cm, width 8cm, height 2cm.
Work out the mass of the gold ingot.
Give your answer in kilograms



$$\text{Density} = \frac{\text{Mass}}{\text{Volume}}$$

$$19.4 = \frac{\text{Mass}}{320}$$

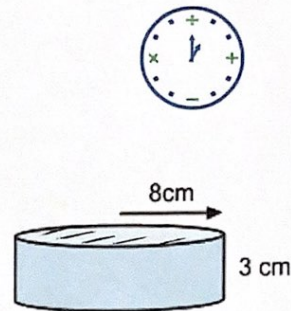
$$19.4 \times 320 = \text{Mass}$$

$$6208 = \text{Mass}$$

$$\text{Mass} = \underline{\underline{6.208\text{kg}}}$$

$$\begin{aligned}\text{Volume} &= \text{Area} \times \text{Depth} \\ &= (8 \times 2) \times 20 \\ &= 320\text{cm}^3\end{aligned}$$

3. A solid cylinder is made of wood.
It has a radius of 8cm and a height of 3cm
The cylinder has a mass of 392 grams
Work out the density of the wood



Give your answer correct to 2 significant figures.

$$\begin{aligned} \text{Density} &= \frac{\text{Mass}}{\text{Volume}} \\ &= \frac{392}{192\pi} \\ &= 0.64988 \text{ g/cm}^3 \\ &= \underline{0.65 \text{ g/cm}^3} \end{aligned}$$

$$\begin{aligned} \text{Volume} &= \text{Area} \times \text{Depth} \\ &= \pi r^2 \times 3 \\ &= \pi 8^2 \times 3 \\ &= 192\pi \\ &= (603.1857895) \end{aligned}$$

4. Seawater has a density of 1.02 g/cm^3 .
255g of seawater mixes with some freshwater, to form coastal water.
The coastal water has a mass of 304.5g, and a density of 1.015 g/cm^3 .
Find the density of the freshwater.

$$\begin{array}{l} \text{Sea} \\ \hline 1.02 \text{ g/cm}^3 \\ \hline 255 \text{g} \end{array}$$

$$\text{Density} = \frac{\text{Mass}}{\text{Volume}} \Rightarrow 1.02 = \frac{255}{\text{Volume}} = 250 \text{ cm}^3$$

$$\begin{array}{l} \text{Fresh} \\ \hline \\ \hline \end{array}$$

$$\text{Density} = \frac{\text{Mass}}{\text{Volume}} \Rightarrow \frac{304.5 - 255}{300 - 250} = \underline{0.99 \text{ g/cm}^3}$$

$$\begin{array}{l} \text{Coast} \\ \hline 1.015 \text{ g/cm}^3 \\ \hline 304.5 \text{g} \end{array}$$

$$\text{Density} = \frac{\text{Mass}}{\text{Volume}} \Rightarrow 1.015 = \frac{304.5}{\text{Volume}} = 300 \text{ cm}^3$$